Antiproliferative Prenylated Stilbenes and Flavonoids from *Macaranga alnifolia* from the Madagascar Rainforest^{#,1}

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Bioassay-guided fractionation of an extract of the fruit of *Macaranga alnifolia* from Madagascar led to the isolation of four new prenylated stilbenes, schweinfurthins E–H (1–4), and one new geranylated dihydroflavonol, alnifoliol (5). The known prenylated stilbene vedelianin (6) and the known geranylated flavonoids bonanniol A (7), diplacol (8), bonannione A (9), and diplacone (10) were also isolated. All 10 compounds were tested for antiproliferative activity in the A2780 human ovarian cancer cell line assay. Vedelianin ($IC_{50} = 0.13 \mu M$) exhibited the greatest activity among all isolates, while schweinfurthin E ($IC_{50} = 0.26 \mu M$) was the most potent of the new compounds.

The genus Macaranga is a large genus of the Euphorbiaceae family. Observation of Macaranga plants in their natural environment has revealed that they produce threadlike wax crystals on their stems, which make the slippery surfaces impassable for all insects except a species of ants known as "wax runners". Chemical analysis has indicated that terpenoids make up a majority of the wax bloom content that helps maintain this symbiotic relationship between plant and insect.² One of the more commonly studied species of this genus is *M. tanarius*, noted for its diterpenoid^{3,4} and flavonoid^{5–7} content. Work has also been performed on the isolation and characterization of terpenes from M. carolinensis,8 flavonoids from M. conifera9 and *M. denticulata*, ¹⁰ chromenoflavones from *M. indica*, ¹¹ clerodane diterpenes from *M. monandra*,¹² bergenin derivatives and polyphenols from *M. peltata*,^{13,14} prenylflavones from *M. pleiostemona*,¹⁵ a geranyl flavanone from *M. schweinfurthii*,¹⁶ tannins from *M.* sinensis,¹⁷ a rotenoid and other compounds from *M. triloba*,¹⁸ and a geranylflavonol from M. vedeliana.¹⁹ No phytochemical studies have been previously reported for M. alnifolia.

Results and Discussion

As part of an ongoing search for cytotoxic natural products from tropical rainforests in Madagascar through the International Cooperative Biodiversity Group (ICBG) program, we obtained an ethanolic extract of the fruit of *Macaranga alnifolia* Baker (Euphorbiaceae) for phytochemical investigation. This extract was found to be active in the A2780 ovarian cancer antiproliferative activity assay, with an IC₅₀ value of 3.5 μ g/mL. Bioassay-guided fractionation led to the isolation of five new compounds: the four new prenylated stilbenes schweinfurthins E–H (1–4) and the new geranylated dihydroflavonol alnifoliol (5). Five known compounds were also isolated: the prenylated stilbene vedelianin (6), the two geranylated dihydroflavonols bonanniol A (7) and diplacol (8), and the two geranylated flavanones bonannione A (9) and diplacone (or nymphaeol A) (10).

Schweinfurthins E–H (1–4) are closely related to schweinfurthins A, B, and D (11, 12, and 14)^{20,21} and vedelianin (6)²² and



are also more distantly related to the prenylated stilbenes schweinfurthin C (13)²⁰ and mappain (15) isolated from *M. mappa*.²³

Schweinfurthin E (1) was isolated as a pale yellow solid with a molecular formula of $C_{30}H_{38}O_6$, based on its HRFABMS. Its UV spectrum, with λ_{max} 331 and 223 nm, correlated well with literature values for compounds of the schweinfurthin class. Its ¹H NMR spectrum indicated the presence of an asymmetrical stilbene core (δ 6.87 ppm, 1H, d, J = 16 Hz, H-1'; δ 6.77 ppm, 1H, d, J = 16.5 Hz, H-2') with both an AA' benzene ring system (δ 6.46 ppm, 2H, s, H-4' and -8') and an AB benzene ring system (δ 6.91 ppm, 1H,

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d, H-6; δ 6.84 ppm, 1H, d, H-8). Proton signals at δ 5.23 (1H, tq, J = 7, 1.5 Hz, H-2"), 3.30 (H-1", partially obscured by solvent), 1.76 (3H, s, H-4"), and 1.65 ppm (3H, s, H-5") indicated the presence of an isoprenyl group. Also present in this spectrum were signals for the protons of three other methyl groups at δ 1.40 (3H, s, H-13), 1.10 (3H, s, H-12), and 1.09 (3H, s, H-11) ppm; protons of a methoxy group at δ 3.84 ppm (3H, s, CH₃O-5); and two methine protons bonded to oxygenated carbons at δ 4.14 (1H, q, J = 3.5, H-3) and 3.30 ppm (H-2, partially obscured by solvent).

The presence of an isoprenyl group was indicated by ¹³C NMR signals at δ 131.1 (C-3"), 124.6 (C-2"), 26.0 (C-5"), 23.3 (C-1"), and 17.9 ppm (C-4"). The three other methyl carbons resonated at δ 29.4 (C-12), 22.0 (C-13), and 16.5 ppm (C-11), and the methoxy carbon resonated at δ 56.5 ppm. Signals for three oxygenated sp³ carbons (C-2, C-4a, and C-3) were present in the spectrum at δ 78.8, 78.1, and 71.8 ppm, respectively, and the carbons of the AA' benzene ring of the stilbene were observed at δ 105.8 ppm for the hydroxylated carbons (C-5' and C-7') and δ 105.8 ppm for the hydrogenated carbons (C-4' and C-8').

The NMR spectra of **1** corresponded closely with those of vedelianin (6)²² and schweinfurthin B (12).²⁰ In particular, the observation of a "quartet" with J = 3.5 Hz for H-3 was in complete agreement with the "quartet" observed for H-3 of schweinfurthin B with J = 3.4 Hz²⁰ and confirmed the *cis* stereochemistry of the C-2 and C-3 hydroxyl groups. The gCOSY, HMBC, and ROESY spectra of **1** (Figure 1 and Supporting Information) and the observed spectroscopic differences between it and the reference compounds were also in complete agreement with this assignment and with the assignment of its structure as 5-*O*-methylvedelianin (or 4"-desisoprenylschweinfurthin B).

Schweinfurthin F (2) was isolated as a pale yellow solid with a molecular formula of $C_{30}H_{38}O_5$, based on HRFABMS. Its ¹H and ¹³C NMR spectra were very similar to those of 1, with the major differences that the NMR signals for H-3 and C-3 were shifted significantly upfield (from δ 4.14 to 2.03 ppm and from δ 71.8 to 39.5 ppm, respectively) when compared to those of 1. These observations, coupled with the fact that the molecule of 2 has five oxygen atoms in place of the six oxygens of 1, suggested that 2 is a 3-deoxy derivative of 1. This was confirmed by the upfield shifts for neighboring hydrogens on the α -side of the molecule (H-4, H-11, H-13) and also for adjacent carbons (C-4, C-11, C-12, C-13). The NMR spectra of 2 were essentially identical with those of a recently prepared synthetic sample,²⁴ thus confirming its structure unambiguously.

Schweinfurthin G (3) was isolated as a pale yellow solid. Its 1 H and 13 C NMR spectra were very similar to those of 2, differing



Figure 1. Key COSY (bold), HMBC (arrows), and ROESY (dashed) correlations of 1.

significantly only in the lack of signals at $\delta \sim 3.8$ and ~ 56 ppm, respectively, corresponding to the methoxy group in **2**. The structure of **3** was thus assigned as 3-deoxyvedelianin.

Schweinfurthin H (4) was isolated as a pale yellow solid. It gave a molecular formula of C₃₀H₃₈O₇, based on HRFABMS, differing from that of 1 by a single oxygen. The ¹H NMR spectrum of 4 indicated the presence of a different asymmetrical stilbene group with a second, alternate AB benzene ring system rather than an AA' benzene ring system; the signals for H-4' (δ 6.52 ppm) and H-8' (δ 6.44 ppm) appeared as two separate peaks. The absence of a ¹H NMR signal for the isoprenyl double bond and upfield shifts of H-2" (δ 3.73 ppm), H-4" (δ 1.33 ppm), and H-5" (δ 1.23 ppm), and the appearance of H-1" as doublet of doublets at δ 2.90 and 2.53 ppm, all indicated that the isoprenyl group was cyclized with one of the phenolic oxygens. The hydroxylation of C-2" was also apparent from its ¹³C NMR chemical shift of δ 70.6 ppm. The final structure was confirmed through NMR comparison with the literature values reported for chiricanine B (16), a tricyclic prenylated stilbene from Lonchocarpus chiricanus.²⁵ The stereochemistry of the 2"-hydroxyl group of 4 was not determined.



The data presented to this point demonstrate the relative stereochemistry of compounds 1-4, but do not establish their absolute stereochemistry. Fortunately this can be established by a comparison of CD spectra and of optical rotations between these compounds and the enantiomers of schweinfurthin F (2). The CD spectra of compounds 1-4 were essentially identical, with strong positive differential dichroic absorptions at 196 nm and strong negative absorptions at 210 nm. Their specific optical rotations at 589 nm were all also similarly positive, with values of +49.2, +50.8, +33.3, and +32.4 for schweinfurthins E–H (1–4). These comparisons establish that compounds 1-4 all belong to the same stereochemical series.

Both the *R*,*R*,*R*- and the *S*,*S*,*S*-enantiomers of schweinfurthin F (2) have recently been synthesized by the Wiemer group.²⁴ They obtained optical rotations of +53.4 for the 1*R*,4a*R*,9a*R* isomer and -55.8 for the 1*S*,4a*S*,9a*S* isomer in CH₃OH; the value for the 1*R*,4a*R*,9a*R* isomer matches well with the value for the natural product (+50.8 in CH₃OH). We conclude that schweinfurthin F (2) has the 1*R*,4a*R*,9a*R* stereochemistry and, thus, that compounds 1, 3, and 4 also have the same 1*R*,4a*R*,9a*R* stereochemistry.

The flavonoid alnifoliol (5) was isolated as a yellow-brown solid with a molecular formula of $C_{25}H_{28}O_7$, based on HRFABMS. The

 Table 1. Antiproliferative Activity of Macaranga alnifolia

 Compounds to A2780 Cells

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compound	IC ₅₀ (µM)
schweinfurthin E (1)	0.26
schweinfurthin $F(2)$	5.0
schweinfurthin G (3)	0.39
schweinfurthin H (4)	4.5
alnifoliol (5)	27.3
vedelianin (6)	0.13
bonanniol A (7)	23.5
diplacol (8)	11.5
bonannione A (9)	24.5
diplacone (10)	10.5

¹H NMR spectrum of **5** showed signals for four aromatic protons (δ 6.81, d, H-2'; δ 6.74, d, H-6'; δ 5.91, s, H-8; δ 5.87, s, H-6), one oxymethine (δ 4.88, d, H-2), and one methine α to a carbonyl (δ 4.47, d, H-3). These data suggested that 5 possesses a dihydroflavanol skeleton. Signals for a geranyl substituent (δ 5.34, m, H-2"; δ 5.10, m, H-7"; δ 3.31, d, H-1"; δ 2.09, td, H-6"; δ 2.02, t, H-5"; δ 1.70, s, H-4"; δ 1.61, s, H-9"; δ 1.56, s, H-10") were also observed. The fact that proton signals for both H-6 and H-8 were present indicated that the geranyl group was on the B-ring. The splitting patterns for H-2' and H-6' confirmed the location of the geranyl group at C-5'. Compound 5 is identical to a known component of propolis, isonymphaeol-B (17), except for the presence of the HO-3 group. Comparison of the NMR spectra of 5 with the literature spectra of 17^{26} fully supported the structural assignment of 5. The coupling constant of the C-2 and C-3 protons (11 Hz) was consistent with their anticoplanar orientation, indicating a trans stereochemistry for the C-2 aryl and C-3 hydroxyl substituents.

The known compounds vedelianin (6),²² bonanniol A (7),²⁷ diplacol (8),^{28,29} bonannione A (9),²⁷ and diplacone (10, also known as nymphaeol A)^{30,31} were also isolated, and their structures were determined on the basis of comparison of their ¹H NMR, ¹³C NMR, and HRFABMS spectra to literature values.^{22,26–31}

All 10 compounds isolated from the fruits of *M. alnifolia* were tested for antiproliferative activity against the A2780 ovarian cancer cell line, and the results are provided in Table 1.

Schweinfurthin E (1) was also tested in the 60-cell human tumor cancer screen at the National Cancer Institute (NCI). The assay measure used by NCI that most closely corresponds to the IC_{50} values for antiproliferative activity is the GI₅₀ value, and schweinfurthin E exhibited a mean panel GI₅₀ of 0.19 μ M. All lines of the leukemia subpanel were found to be highly sensitive to 1, while all lines of the ovarian cancer subpanel (OVCAR-3, -4, and -8 and SK-OV-3) were somewhat resistant, with GI₅₀ values averaging $2.2 \,\mu$ M. This result is somewhat surprising in view of the sensitivity of the A2780 ovarian cancer cell line to 1, but can be explained in part by the fact that the A2780 cell line is a drug-sensitive line.³² The most sensitive lines included leukemia (MOLT-4) and CNS (SF-295) and renal (A498 and CAKI-1) cancers, which all gave GI_{50} and TGI values of <10 nM. The complete mean graph for 1 is provided as Supporting Information to this article. These differential cytotoxicity results suggest that schweinfurthin E(1), similar to the other schweinfurthins, may share a similar mechanism of action with the stelletins.

It is instructive to compare the data reported above with the previously reported data for schweinfurthins $A-D^{20,21}$ and for 3-deoxyschweinfurthin and its synthetic analogues.^{24,33} The literature data were reported primarily as mean GI₅₀ values from the NCI 60-cell line screen, and the data above are only for one cell line, so comparisons are only possible between compounds determined in the same bioassay.

The first comparisons are between the cytotoxicities of schweinfurthin E (1, IC₅₀ 0.26 μ M) and vedelianin (6, IC₅₀ 0.13 μ M) and those of schweinfurthin G (3, IC₅₀ 0.39 μ M) and vedelianin. In the first case, vedelianin is twice as potent as schweinfurthin E, suggesting that the replacement of the C-5 hydroxyl group with a methoxy group is deleterious to activity. In the second case, vedelianin (**6**) is about 3 times as potent as schweinfurthin G (**3**), indicating that the C-3 hydroxyl group enhances activity in this series. However, the issue is not as simple as this, because 3-deoxyschweinfurthin B is slightly more active than schweinfurthin B,³⁴ so clearly the length of the side chain has an influence on how the C-3 hydroxyl group affects activity.

Another direct comparison is possible between the mean GI₅₀ values of schweinfurthin E (1, 0.19 μ M) and schweinfurthin B (12, 0.79 μ M); this indicates that the shorter geranyl side chain of 1 enhances its activity. A final comparison between schweinfurthin F (2, IC₅₀ 5.0 μ M) and schweinfurthin G (3, IC₅₀ 0.39 μ M) indicates that the combination of the loss of the C-3 hydroxyl group with methylation of the C-5 hydroxyl group results in a greater loss of activity than would have been predicted by either modification alone. The trend in all these comparisons is for the more polar compound to be more active, so it is possible that the observed activity is limited in some way by aqueous solubility, but further experiments are required to confirm this suggestion. Schweinfurthin C (13) was found to be much less active than schweinfurthins A, B, and D, so cyclization of the geranyl group must play an important role in mediating the biological activity of these compounds.

Experimental Section

General Experimental Procedures. Optical rotations were recorded on a Perkin-Elmer 241 polarimeter. CD spectra were recorded on a JASCO J-700 spectrometer. NMR spectra were obtained on a JEOL Eclipse 500 or a Varian INOVA 400 MHz spectrometer. The chemical shifts are given in δ (ppm), and coupling constants are reported in Hz. FAB mass spectra were obtained on a JEOL JMS-HX-110 instrument. HPLC was performed on a Shimadzu LC-10AT instrument with a semipreparative C₈ Varian Dynamax column (5 μ m, 250 \times 10 mm) and a preparative phenyl Varian Dynamax column (8 μ m, 250 \times 21.4 mm). Finnigan LTQ LC/MS with a C₁₈ Hypersil column (5 μ m, 100 \times 2.1 mm) was also used for crude sample analysis.

Antiproliferative Bioassays. Antiproliferative activity measurements were performed at Virginia Polytechnic Institute and State University against the A2780 ovarian cancer cell line as previously described. The A2780 cell line is a drug-sensitive human ovarian cancer cell line.³²

Plant Material. Immature and mature fruits of *Macaranga alnifolia* Baker (Euphorbiaceae) (vernacular name "Mokaranana") were collected in November 2001. The specimens were collected around the Natural Reserve of Zahamena in the province of Toamasina, Madagascar, at coordinates 17-41-01 S and 48-38-28 E, at an elevation of 900 m. Duplicate voucher specimens have been deposited at the Centre National d'Application des Recherches Pharmaceutiques (CNARP) and the Direction des Recherches Forestieres et Piscicoles Herbarium (TEF) in Antananarivo, Madagascar; the Missouri Botanical Garden in St. Louis, Missouri (MO); and the Museum National d'Histoire Naturelle in Paris, France (P).

Extraction and Isolation. Dried fruits of M. alnifolia (275 g) were ground in a hammer mill, then extracted with EtOH by percolation for 24 h at rt to give the crude extract MG 1021 (12.8 g), of which 2.84 g was made available to Virginia Polytechnic Institute and State University. The crude bioactive extract MG 1021 (IC₅₀ = $3.5 \mu g/mL$, 2.32 g) was partitioned between hexanes (200 mL) and MeOH-H₂O (4:1, 200 mL). The aqueous fraction was dried and subsequently partitioned between 1-BuOH and H2O. The evaporated 1-BuOH fraction (1.96 g) displayed antiproliferative activity (IC₅₀ = 1.0 μ g/mL) and was further separated by repeated RP-C₁₈ column chromatography. The fractions eluted with 70% and 80% $MeOH{-}H_2O$ showed the most improved activity and were separated by solid-phase extraction into fractions eluting with MeOH-H2O (3:2) and MeOH. Preparative RP-C18 HPLC using MeOH-H2O (4:1, 1 mL/min) on these bioactive eluates and combination of similar fractions yielded a total of 16 new fractions (A-K and L-P). Fraction D was identified as schweinfurthin E (1, $t_{\rm R}$ 21.5 min, 25.4 mg), while fractions A–C yielded vedelianin (6, t_R 17.1 min, 4.1 mg), schweinfurthin G (3, t_R 18.2 min, 0.9 mg), and schweinfurthin H (4, t_R 19.5 min, 1.5 mg), respectively, upon additional purification by semipreparative RP-C₁₈ and RP-phenyl HPLC, eluting with MeOH-H₂O, 4:1. Fraction F was also identified as schweinfurthin F (**2**, *t*_R 25.9 min, 10.6 mg). Fractions G (*t*_R 32.6 min) and H (*t*_R 30-45 min) were combined and purified by semipreparative RP-phenyl HPLC to obtain both alnifoliol (**5**, 24.9 mg) and diplacone (**10**, 34.1 mg). Additionally, fractions M, N, and P yielded diplacol (**8**, *t*_R 19 min, 6.7 mg), bonanniol A (**7**, *t*_R 21 min, 27.1 mg), and bonannione A (**9**, *t*_R 35 min, 3.0 mg). The structures of the known compounds were identified by comparison of their spectroscopic data with literature values.^{22,27-31}

Schweinfurthin E (1): pale yellow solid; $[\alpha]^{22}_{D}$ +49.2 (c 0.13, MeOH); UV (MeOH) λ_{max} (log ϵ) 223 (4.5), 331 (4.5) nm; CD (MeOH) λ_{max} ($\Delta \epsilon$, dm³ mol⁻¹ cm⁻¹) 196 (+173), 210 (-139), 222 (-135), 250 (+11); ¹H NMR (CD₃OD, 500 MHz) δ 6.91 (1H, d, J = 1.5 Hz, H-6), 6.87 (1H, d, J = 16 Hz, H-1'), 6.84 (1H, d, H-8), 6.77 (1H, d, J =16.5 Hz, H-2'), 6.46 (2H, s, H-4', 8'), 5.23 (1H, tq, J = 7, 1.5 Hz, H-2"), 4.14 (1H, q, J = 3.5 Hz, H-3), 3.84 (3H, s, CH₃O-5), 3.30 (partially obscured by solvent, H-2, 1"), 2.76 (2H, m, H-9), 2.34 (1H, dd, J = 14, 3 Hz, H-4), 1.93 (1H, dd, J = 13.5, 3.5 Hz, H-4), 1.76 (3H, s, H-4"), 1.74 (1H, dd, J = 12.5, 6 Hz, H-9a), 1.65 (3H, s, H-5"), 1.40 (3H, s, H-13), 1.10 (3H, s, H-12), 1.09 (3H, s, H-11); ¹³C NMR (CD₃OD, 125 MHz) δ 157.3 (C-5', 7'), 150.2 (C-5), 143.4 (C-10a), 137.6 (C-3'), 131.1 (C-3"), 130.8 (C-7), 128.6 (C-1'), 127.7 (C-2'), 124.6 (C-2"), 124.4 (C-8a), 121.7 (C-8), 116.0 (C-6'), 108.3 (C-6), 105.8 (C-4', 8'), 78.8 (C-2), 78.1 (C-4a), 71.8 (C-3), 56.5 (CH₃O-5), 48.5 (C-9a) 44.8 (C-4), 39.2 (C-1), 29.4 (C-12), 26.0 (C-5"), 24.0 (C-9), 23.3 (C-1"), 22.0 (C-13), 17.9 (C-4"), 16.5 (C-11); HRFABMS m/z 494.2646 [M]⁺ (calcd for C₃₀H₃₈O₆, 494.2668).

Schweinfurthin F (2): pale yellow solid; $[\alpha]^{22}_{D}$ +50.8 (c 0.06, MeOH); UV (MeOH) λ_{max} (log ϵ) 224 (4.4), 331 (4.4) nm; CD (MeOH) λ_{max} ($\Delta\epsilon$, dm³ mol⁻¹ cm⁻¹) 196 (+176), 210 (-134), 220inf (-130) 247 (+9.5), 258 (+16); ¹H NMR (CD₃OD, 500 MHz) δ 6.91 (1H, d, J = 1.5 Hz, H-6), 6.86 (1H, d, J = 16.5 Hz, H-1'), 6.83 (1H, d, J =1.5 Hz, H-8), 6.77 (1H, d, J = 16.5 Hz, H-2'), 6.46 (2H, s, H-4', 8'), 5.23 (1H, tq, J = 7, 1.5 Hz, H-2"), 3.83 (3H, s, CH₃O-5), 3.30 (partially obscured by solvent, H-2, 1"), 2.72 (2H, m, H-9), 2.03 (2H, m, H-3), 1.79 (1H, m, H-4), 1.76 (3H, s, H-4"), 1.75 (1H, m, H-9a), 1.65 (1H, m, H-4), 1.65 (3H, s, H-5"), 1.21 (3H, s, H-13), 1.09 (3H, s, H-12), 0.87 (3H, s, H-11); ¹³C NMR (CD₃OD, 125 MHz) δ 157.3 (C-5', 7'), 150.2 (C-5), 143.7 (C-10a), 137.6 (C-3'), 131.2 (C-3"), 130.9 (C-7), 128.6 (C-1'), 127.8 (C-2'), 124.6 (C-2"), 124.1 (C-8a), 121.8 (C-8), 116.0 (C-6'), 108.3 (C-6), 105.8 (C-4', 8'), 78.8 (C-2), 78.2 (C-4a), 56.5 (CH₃O-5), 39.5 (C-3), 39.0 (C-1), 29.0 (C-4), 27.9 (C-12), 26.0 (C-5"), 24.1 (C-9), 23.3 (C-1"), 20.2 (C-13), 17.9 (C-4"), 14.9 (C-11); HRFABMS m/z 478.2737 [M]⁺ (calcd for C₃₀H₃₈O₅, 478.2719).

Schweinfurthin G (3): pale yellow solid; $[\alpha]^{22}D$ +33.3 (c 0.03, MeOH); UV (MeOH) λ_{max} (log ϵ) 228 (4.2) 329 (3.9) nm; CD (MeOH) λ_{max} ($\Delta\epsilon$, dm³ mol⁻¹ cm⁻¹) 196 (+177), 210 (-133), 220 (-123), 247 (+10); ¹H NMR (CD₃OD, 500 MHz) δ 6.80 (1H, d, J = 17 Hz, H-1'), 6.79 (1H, d, H-6), 6.72 (1H, d, J = 1.5 Hz, H-8), 6.70 (1H, J = 16 Hz, H-2'), 6.44 (2H, s, H-4', 8'), 5.23 (1H, tq, J = 7, 1.5 Hz, H-2"), 3.30 (partially obscured by solvent, H-2, 1"), 2.71 (2H, m, H-9), 2.06 (2H, m, H-3), 1.80 (1H, m, H-4), 1.76 (3H, s, H-4"), 1.75 (1H, m, H-9a), 1.68 (1H, m, H-4), 1.65 (3H, s, H-5"), 1.23 (3H, s, H-13), 1.10 (3H, s, H-12), 0.88 (3H, s, H-11); ¹³C NMR (CD₃OD, 125 MHz) δ 157.3 (C-5', 7'), 147.0 (C-5), 142.2 (C-10a), 131.3 (C-3"), 137.6 (C-3'), 131.0 (C-7), 128.6 (C-1'), 127.5 (C-2'), 124.6 (C-2"), 124.0 (C-8a), 120.4 (C-8), 115.9 (C-6'), 111.1 (C-6), 105.7 (C-4', 8'), 78.8 (C-2), 78.2 (C-4a), 39.5 (C-3), 38.9 (C-1), 29.0 (C-4), 27.9 (C-12), 26.0 (C-5"), 24.0 (C-9), 23.3 (C-1"), 20.3 (C-13), 17.9 (C-4"), 14.8 (C-11). HRFABMS m/z 464.2595 [M]⁺ (calcd for C₂₉H₃₆O₅, 464.2563).

Schweinfurthin H (4): pale yellow solid; $[α]^{22}_{D} + 32.4$ (*c* 0.04, MeOH); UV (MeOH) λ_{max} (log ϵ) 224 (4.5), 329 (4.4) nm; CD (MeOH) λ_{max} ($\Delta \epsilon$, dm³ mol⁻¹ cm⁻¹) 196 (+232), 210 (-145), 220inf (-135), 250 (+12); ¹H NMR (CD₃OD, 500 MHz) δ 6.93 (1H, d, *J* = 1.5 Hz, H-6), 6.90 (1H, d, *J* = 16 Hz, H-1'), 6.85 (1H, d, *J* = 1 Hz, H-8), 6.80 (1H, d, *J* = 1 16 Hz, H-2'), 6.52 (1H, d, *J* = 1.5 Hz, H-4'), 6.44 (1H, d, *J* = 1 Hz, H-8'), 4.14 (1H, q, *J* = 3.5 Hz, H-3), 3.84 (3H, s, CH₃O-5), 3.73 (1H, dd, *J* = 7.5, 5.5 Hz, H-2''), 3.30 (1H, m, H-2), 2.90 (1H, dd, *J* = 17, 5.5 Hz, H-1''), 2.76 (2H, m, H-9), 2.53 (1H, dd, *J* = 17, 7.5 Hz, H-1''), 2.34 (1H, dd, *J* = 14, 3 Hz, H-4), 1.92 (1H, dd, *J* = 14.5 Hz, H-4), 1.74 (1H, dd, *J* = 12, 5.5 Hz, H-9a), 1.40 (3H, s, H-13), 1.33 (3H, s, H-4''), 1.23 (3H, s, H-5''), 1.10 (3H, s, H-12), 1.09 (3H, s, H-11); ¹³C NMR (CD₃OD, 125 MHz) δ 157.1 (C-5'), 155.3 (C-7'), 150.2 (C-5), 143.5 (C-10a), 138.5 (C-3'), 130.6 (C-7), 129.1 (C-1'),

127.5 (C-2'), 124.4 (C-8a), 121.9 (C-8), 108.4 (C-4'), 108.4 (C-6), 107.6 (C-6'), 105.0 (C-8'), 78.8 (C-2), 78.1 (C-4a), 77.7 (C-3''), 71.8 (C-3), 70.6 (C-2''), 56.5 (CH₃O-5), 44.8 (C-4), 39.2 (C-1), 29.4 (C-12), 27.4 (C-1''), 25.8 (C-5''), 24.0 (C-9), 22.0 (C-13), 20.8 (C-4''), 16.6 (C-11); HRFABMS m/z 510.2579 [M]⁺ (calcd for C₃₀H₃₈O₇, 510.2618).

Alnifoliol (5): yellowish-brown solid; $[\alpha]^{23}_{D} + 15.3$ (*c* 0.25, MeOH); UV (MeOH) λ_{max} (log ϵ) 213 (4.7), 290 (4.4) nm; CD (MeOH) λ_{max} ($\Delta\epsilon$, dm³ mol⁻¹ cm⁻¹) 196 (+150), 210 (-123), 220inf (-116), 262 (+16), 295 (-19); ¹H NMR (CD₃OD, 500 MHz) δ 6.81 (1H, d, H-2'), 6.74 (1H, d, J = 2 Hz, H-6'), 5.91 (1H, d, J = 2.5 Hz, H-8), 5.87 (1H, d, H-6), 5.34 (2H, t, H-2''), 5.10 (2H, t, H-7''), 4.88 (1H, d, H-2), 4.47 (1H, d, J = 11 Hz, H-3), 3.31 (2H, d, J = 7.5 Hz, H-1''), 2.09 (2H, q, J = 7.5 Hz, H-6''), 2.02 (2H, t, J = 8 Hz, H-5''), 1.70 (3H, s, H-4''), 1.61 (3H, s, H-9''), 1.56 (3H, s, H-10''); ¹³C NMR (CD₃OD, 125 MHz) δ 197.0 (C-4), 167.4 (C-7), 164.0 (C-5), 163.2 (C-9), 144.5 (C-3'), 143.6 (C-4'), 135.5 (C-3''), 130.9 (C-8''), 128.1 (C-1'), 127.6 (C-5'), 124.1 (C-7'), 122.5 (C-2''), 120.0 (C-6'), 111.8 (C-2'), 100.5 (C-10), 96.0 (C-8), 95.0 (C-6), 84.1 (C-2), 72.4 (C-3), 39.6 (C-5''), 27.8 (C-1''), 26.4 (C-6''), 24.6 (C-9''), 16.4 (C-10''), 14.9 (C-4''); HRFABMS m/z 440.1831 [M]⁺ (calcd for C₂₅H₂₈O₇, 440.1835).

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Supporting Information Available: Characterization data for compounds 6–10, NCI 60-cell line data for 1, ¹H and ¹³C NMR spectra for compounds 1–10, and gCOSY, HMQC, HMBC, and ROESY spectra of compound 1. The plots of the ¹³C NMR spectra are offset, and the correct values are approximately 1.5 ppm higher than indicated. This material is available free of charge via the Internet at http:// pubs.acs.org.

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